Influences of aerosol physiochemical properties and new particle formation on CCN activity from observation at a suburban site of China

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A B S T R A C T

With the aim of understanding the impact of aerosol particle size and chemical composition on CCN activity, the size-resolved cloud condensation nuclei (CCN) number concentration (N_{CCN}), particle number size distribution (PSD) (10–600 nm), and bulk chemical composition of particles with a diameter < 1.0 μm (PM_{1}) were measured simultaneously at Xinzhou, a suburban site in northern China, from 22 July to 26 August 2014. The N_{CCN} was measured at five different supersaturations (SS) ranging from 0.075%–0.76%, Diurnal variations in the aerosol number concentration (N_{a}), N_{CCN}, the bulk aerosol activation ratio (AR), the hygroscopicity parameter (κ_{chem}), and the ratio of 44 mass to charge ratio (m/z 44) to total organic signal in the component spectrum (f_{44}) were examined integrally to study the influence of particle size and chemical composition on CCN activation. We found that particle size was more related to the CCN activation ratios in the morning, whereas in the afternoon (~1400 LST), N_{chem} and f_{44} were more closely associated with the bulk AR. Assuming the internal mixing of aerosol particles, N_{CCN} was estimated using the bulk chemical composition and real-time PSD. We found that the predicted CCN number concentrations were underestimated by 20–30% at SS < 0.2% due to the measurement uncertainties. Estimates were more accurate at higher SS levels, suggesting that the hygroscopicity parameter based on bulk chemical composition information can provide a good estimate of CCN number concentrations. We studied the impacts of new particle formation (NPF) events on size-resolved CCN activity at the “growth” stage and “leveling-off” stage during a typical NPF event by comparing with the case during non-NPF event. It has been found that CCN activation was restrained at the “growth” stage during which larger particle diameters were needed to reach an activation diameter (D_{a}) and the bulk AR decreased as well. However, during the “leveling-off” stage, a lower D_{a} was observed and CCN activation was greatly enhanced.

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1. Introduction

As a subset of atmospheric aerosols, cloud condensation nuclei (CCN) enable the condensation of water vapor and the formation of cloud droplets. When the water vapor content of a cloud remains constant, elevated CCN number concentrations increases the number of cloud droplets but decreases their sizes. This changes the cloud albedo (Twomey, 1977), the cloud life span (Albrecht, 1989), cloud microphysical properties, and ultimately, climate. Evaluating the interaction between aerosol particles and clouds is one of the largest uncertainties in the quantification of aerosol effects on climate (IPCC, 2013; Chin et al., 2009) with CCN at the heart of the problem (Medina et al., 2007). Modelers have simulated the global spatial distribution of CCN concentration (Lee et al., 2012), but with large uncertainty. Thus, obtaining an accurate temporal and spatial CCN number distribution is an important and challenging issue. Many ground-based CCN observation field campaigns have been performed at different sites with different surface types around the world, such as those on mountains (DeFelice and Saxena, 1994; Miao et al., 2015), in forests (Gunthe et al., 2009), on plains (Rissman et al., 2006; Deng et al., 2011, 2013),
over oceans (Yum and James, 2004) and in megacities (Almeida et al., 2014; Leng et al., 2014a, 2014b). Owing to the complicated properties and short life span of aerosols, CCN characteristics can fluctuate over time at a fixed site (e.g. Almeida et al., 2014; Zhang et al., 2014), while direct long-term measurements of CCN number concentrations on a global scale is likely unrealistic. It is key to understand what controls the ability of aerosols to act as CCN and to develop typical CCN parameterization schemes for use in models. Aerosol activation depends on the aerosol mixing state, size distribution and chemical composition, however, there is no consensus as to how accurate is the measurement for each of these factors is needed to predict CCN number concentrations (Dusek et al., 2006; Rissman et al., 2006; McFiggans et al., 2006; Anttila and Kerminen, 2007; Hudson, 2007; Quinn et al., 2008; Zhang et al., 2012; Ma et al., 2013). Aerosol-CCN closure study, or comparing predicted CCN concentrations with measurements based on Köhler theory, is a valid approach to test the relative significance of aerosol physical and chemical characteristics. It can also help to form a framework for predicting $N_{CCN}$.

New particle formation (NPF), recognized as contributing toward 40%–55% of the total aerosol number concentration (Yu et al., 2008), was observed to enhance $N_{CCN}$ in both a relatively clean environment (O’Dowd, 2001; Sihto et al., 2011; Kalkavouras et al., 2017) and at polluted sites (Wiedensohler et al., 2009; Yue et al., 2011; Ma et al., 2016). For example, Kuang et al. (2009) developed a semi-analytical model for CCN prediction by simulating the growth of a subset of an aerosol population from 3 to 100 nm and found that the NPF enhanced CCN number concentrations by a mean factor of 3.8. Asmi et al. (2011) assumed that particles with sizes larger than 80 nm would activate into CCN. On average, NPF increased CCN number concentrations by a factor of 2.11 ± 1.14. To our knowledge, most of the previous studies have not yet investigated the impacts on size-resolved CCN activation ratio as well as changes in critical diameter during NPF events. The goal of our research is to investigate the variation in the critical diameter of particles in an NPF case against that in a background case using size-resolved CCN observations in an attempt to study the influence of NPF on CCN activity. In particular, we will focus on the “growth” stage and the “leveling-off” stage of the NPF to investigate the effect with the aim of improving the parameterization of CCN in climate models.

In China, the high number concentration and complex properties of aerosols from both the ground and vertical profile measurements (Li et al., 2007; Li et al., 2015) increases the variability in cloud droplet formation, which has a profound effect on climate change. Supported by the National Project “Observation and Modelling Study of Cloud, Aerosol and their Climate Effect”, we conducted a field campaign called “Atmosphere, Aerosol, Cloud and CCN” ($A2C2$) from 22 July to 26 August 2014 in northern China, which aimed to understand the impact of aerosol particle size and chemical composition on CCN activity. In the paper, Section 2 describes the observation site and data used in the study. Section 3 describes the theory and methods, including derivations of the hygroscopic parameter and critical diameter ($D_{crit}$). Section 4 presents the overall characteristics of aerosols, CCN number concentrations, aerosol activation, the CCN closure study using bulk chemical composition information and the impact of NPF on CCN activity. Conclusions were summarized in the last section.

2. Observation site and measurements

2.1. Observation site

The observation site, Xinzhou (38.24°N, 112.43°E; 1500 m asl XZ), is located on the border of the Loess Plateau and the Northern China Plain, about 78 km northwest of Taiyuan (Fig. 1). The site experiences a warm temperate continental monsoon climate. During the observation period, the Xinzhou site was surrounded by fields of corn plants. In summer, the dominant winds come from the northeast and southwest. The SMPS combined with the DMT CCN–100 was set during the campaign from 22 July to 26 August 2014. Details about the site information and measurement can also be found in Zhang et al. (2016).

2.2. Measurements

The instruments were housed in a $2 \times 6 \times 2$ m container, surrounded by corn plant fields. The automatic meteorological station that measured the ambient temperature, relative humidity, rain rate, wind speed, and wind direction was about 20 m away from the container. Atmospheric aerosol particles were collected from a sampling inlet that was installed –1.5 m above the roof of the container, then passed through a silica gel desiccant drying tube, which assured that the relative humidity level was below 30% (Leng et al., 2013), before entering into a TSI Scanning Mobility Particle Sizer (SMPS). The SMPS is comprised of a 3080 electric classifier, a 3022A condensation particle counter (CPC), and a microcomputer. The particle number size distribution

Fig. 1. Geographic locations of Xinzhou site.
for the mobility diameter was between 10 and 600 nm. Because we measured the size-resolved CCN number concentrations, which are with the particle size range of 10–600 nm and the large particles were included here. Therefore, without considering of the large particles will not cause an uncertainty in CCN closure study. A set of particle number size distributions was obtained every 255 s: 240 s for the voltage upscan time and 15 s retrace time. A Droplet Measurement Technology Cloud Condensation Nuclei counter (CCNC; Roberts and Nenes, 2006) was situated downstream of the SMPS. The whole experimental setup and method is also described as Scanning Mobility CCN Analysis (SMCA), introduced by Moore et al. (2010). The atmospheric aerosol flow streaming out of the differential mobility analyzer (DMA), which is a basic part the electric classifier, at a rate of 0.8 l per minute (lpm) was split, going into the CPC at a flowrate of 0.3 lpm and into the CCNC with a sheath-to-sample flowrate ratio of 10. Five SS (0.07%, 0.1%, 0.2%, 0.4%, and 0.8%, each running for 10 min), were set in the CCNC and corrected to 0.075%, 0.13%, 0.17%, 0.39%, and 0.76% after being calibrated with pure ammonium sulfate following the procedures described by Rose et al. (2008). Thus, two sets of particle number size distribution for each SS were obtained. The first five minutes of data were removed to exclude the influence of the transition in SS according to Deng et al. (2011). The fluctuation (relative standard deviation) for the second five minutes bulk CCN data is <3%. The SMPS and CCNC data were processed with multiple corrections following the method developed by Deng et al. (2011).

Black carbon (BC) was measured with a 7-wavelength online aethalometer (AE-31, Magee Scientific), whose flow rate setting was 5 lpm. The time resolution of BC data was 5 min. By comparing the difference between light transmission through the particle-laden sample spot and the particle-free reference spot in the filter, the BC concentration can be obtained (Cheng et al., 2006; Dumka et al., 2010). The bulk chemical composition of particles with a diameter <1.0 μm (PM1) was measured with an Aerodyne Aerosol Chemical Speciation Monitor (ACSM) at a time resolution of ~7.5 min. By this technique, the non-refractory submicron aerosol species including organics, sulfate, nitrate, ammonium, and chloride was measured simultaneously with the CCN measurement. Before sampling into the ACSM, aerosol particles were dried using a silica gel desiccant. The ACSM was operated with a scan rate of mass spectrometer at 500 ms amu⁻¹ from m/z 10 to 150. $f_{org}$ is calculated by the mass signal at m/z 44, which refers to the fraction of mass signal at m/z 44 in total organics. Higher $f_{org}$ in organics indicates the particles being more oxidized. Regarding the calibration of the ACSM, mono-dispersed, size-selected 300 nm ammonium nitrate particles within a range of concentrations were sampled into both the ACSM and a condensation particle counter (CPC). The ionization efficiency (IE) was then determined by comparing the response factors of the ACSM to the mass calculated with known particle size and number concentrations from the CPC. A collection efficiency factor of 0.5 was applied to account for the incomplete detection of particles primarily due to particle bounce effect, and was found to be appropriate for this site. A comparison of particle mass concentration from the ACSM with the integrated mass concentration from the SMPS was also made. The number concentrations measured by the SMPS were converted to the mass concentrations using chemically-resolved particle density that was estimated from the chemical composition of PM1. The result showed that the PM1 (total mass of non-refractory and black carbon) from ACSM was well correlated with that of SMPS (with $r^2 = 0.65$), and the mass ratio of SMPS to ACSM is 0.62. A more detailed diagnostics of CE and the inter-comparison with SMPS measurements is given in Wang et al. (2016). And more detailed description of the operation and calibration of the ACSM are given by Ng et al. (2011) and Sun et al. (2012). In this study, the refractory components of aerosol include such as black carbon and dust. For black carbon, it accounts for about 2.4% of total PM1; for dust, because of their large size (usually in coarse mode) and thus is not within the detect range of the ACSM technique.

3. Derivation of hygroscopic parameter $\kappa_{chem}$ and critical dry diameter $D_{crit}$

In this study, we calculated $\kappa_{chem}$ based on bulk chemical composition observations during the field campaign. Assuming an internal mixture of chemical species as measured by the ACSM, the hygroscopic parameter $\kappa_{chem}$ was derived according to the equation, $K = \sum_i c_i n_i$, where $c_i$ and $n_i$ are the volume fraction and hygroscopic parameter, respectively, of the ith species (Petters and Kreidenweis, 2007).

According to ACSM measurements, chemical composition constituents at the Xinzhou site were mainly organics, sulfate, and nitrate. The volume fraction of chloride and BC was <5%. Therefore, the particle hygroscopic parameter is (Petters and Kreidenweis, 2007):

$$K_{chem} = K_{Org} E_{Org} + K_{(NH_4)_2SO_4} E_{(NH_4)_2SO_4} + K_{NH_4NO_3} E_{NH_4NO_3}.$$

The hygroscopicity parameters for $(NH_4)_2SO_4$ and NH4NO3 are 0.61 and 0.67 (Petters and Kreidenweis, 2007). The value of $\kappa_{org}$ during the
field campaign can be estimated as: \( \kappa_{\text{org}} = 2.10 \times f_{44} - 0.11 \), where \( f_{44} \) is the fraction of m/z 44 in total organics (Mei et al., 2013). The volume fraction of each chemical species was obtained by dividing its mass concentration by its density. The densities of organics, \((\text{NH}_4)_2\text{SO}_4\), and \(\text{NH}_4\text{NO}_3\) used in this work are 1200 kg m\(^{-3}\) (Turpin and Lim, 2001), 1770 kg m\(^{-3}\) and 1720 kg m\(^{-3}\) (Larson et al., 1988), respectively. The mean chemical composition derived \( \kappa_{\text{chem}} \) value during the entire period was 0.42 ± 0.04. This value is higher than the global mean \( \kappa \) value for continental regions (0.27 ± 0.21) and much lower than that for marine regions (0.72 ± 0.24) (Andreae and Rosenfeld, 2008; Pringle et al., 2010).

Then we retrieved the critical diameter \( (D_{\text{crit}}) \) based on a \( \kappa \)-Köhler equation (Petters and Kreidenweis, 2007), which is described as,

\[
S = \left( 1 + \kappa \frac{D_0^3}{D^3 - D_0^3} \right)^{-1} \exp \left( \frac{4\sigma M_w}{RT \rho_w D} \right),
\]

where \( \kappa \) is hygroscopic parameter, which is defined through its effect on the water activity of the solution; \( S \) is the supersaturation, \( D \) is the droplet diameter under different \( S \), \( D_0 \) is the dry diameter of the particle, \( M_w \) is the molecular weight of water, \( \sigma \) is the surface tension of pure water.

Fig. 3. From top to bottom: time series of the aerosol particle number size distribution; integral\((10–600 \text{ nm, blue curve})\) and nucleation-mode\((10–30 \text{ nm, red curve})\) CN number concentrations; CCN number concentrations at supersaturations of 0.075% (black dots), 0.13% (red dots), 0.17% (blue dots), 0.39% (green dots), and 0.76% (cyan dots); bulk CCN activation ratios at supersaturations of 0.075% (black dots), 0.13% (red dots), 0.17% (blue dots), 0.39% (green dots), and 0.76% (cyan dots). The time period is 22 July to 26 August 2014.
\( \rho_w \) is the density of water, \( R \) is the gas constant, and \( T \) is the absolute temperature.

When \( \kappa > 0.2 \), it can be calculated from the following approximate expression:

\[
\kappa = \frac{4A^3}{27D^3 \ln^2 S}
\]

\[ A = \frac{4\sigma M_w}{RT\rho_w} \]

Here the \( \kappa \) equal to \( \kappa_{\text{chem}} \), we thus calculated the critical dry diameter \( (D_{\text{crit}}) \). The \( D_{\text{crit}} \) of a particle with properties \( (S, \kappa_{\text{chem}}) \) corresponds to the minimum value of \( D \) needed for the particle to be activated. Particles with sizes larger than \( D_{\text{crit}} \) will be activated.

4. Results

4.1. Overview of meteorological parameters, aerosol and CCN number concentrations during the field campaign

During the observation period, the average temperature was 21.6 °C and the average ambient relative humidity was 72.3%. Fig. 2 shows that surface temperature and relative humidity have opposite trends. Five short-term precipitation events occurred during the field campaign. The largest rain rate was \(< 10 \text{ mm h}^{-1} \). When it rained, aerosol number concentrations dropped sharply. The dominant winds at Xinzhou were from the northeast and southwest during the field campaign.

Clear days were common at Xinzhou with many episodes of gas-to-particle formation (Shen et al., 2011). An NPF event was identified if (1) there was a sudden burst of nucleation mode particles and (2) the formation of nucleation mode particles lasted for several hours with a

![Diagram](image-url)
The significant growth to larger particles, which appears as a “banana” shape in the particle number size distribution and is representative of a NPF event (Shi et al., 2001; Heintzenberg et al., 2007; Olofson et al., 2009; Wang et al., 2013), was only observed on a few days. Most of the cases merely showed a sudden increase in aerosol number concentration but without growing to larger particles (Fig. 3, the first panel from the top), which we call non-typical NPF. The particle size diameter \( D \) corresponding to the maximum \( dN/d\log D \) is \( \sim 100 \) nm, of which 60% of particles were in the nucleation mode.

According to Fig. 3 (the second and third panel from the top), both the aerosol and CCN number concentrations fluctuated largely during the observation period. The average aerosol number concentration is \( 12,195 \pm 5,657 \) cm\(^{-3} \), among which particles with \( D \) ranging from 10 to 30 nm accounted for 32.7% of all particles. Measured mean CCN number concentrations are \( 13,533 \pm 7,777 \) cm\(^{-3} \), \( 17,699 \pm 9,747 \) cm\(^{-3} \), \( 32,640 \pm 15,999 \) cm\(^{-3} \), \( 50,574 \pm 23,255 \) cm\(^{-3} \), \( 69,213 \pm 33,030 \) cm\(^{-3} \) at \( SS \) of 0.075%, 0.13%, 0.17%, 0.39%, and 0.76%, respectively. Both the mean values and standard deviations increased with increasing \( SS \). According to last panel in Fig. 3, the average bulk activation ratio, the ratio of \( N_{\text{CCN}} \) to \( N_{\text{CN}} \), increased with increasing \( SS \), i.e., 12.6 ± 8%, 16.4 ± 9%, 29.4 ± 13%, 43.7 ± 15%, and 58.1 ± 15% at \( SS \) of 0.075%, 0.13%, 0.17%, 0.39%, and 0.76%, respectively. Both the mean values and standard deviations increased with increasing \( SS \).

4.2. Diurnal variations

To investigate the influence of particle size and chemical composition on CCN activation, diurnal variations in \( N_{\text{CN}} \), \( N_{\text{CCN}} \), bulk AR, \( \kappa_{\text{chem}} \), and \( f_{44} \) and the particle number size distribution over all days were examined (Fig. 4). Here, the \( f_{44} \) refers to the fraction of total organic mass signal at \( m/z \) 44. The \( m/z \) 44 signal is mostly due to acids or acid-derived species, such as esters, and \( f_{44} \) is closely related to the organic oxidation level, i.e., O:C ratio. Oxidized/aged acids are generally more hygroscopic and easily activated. Both CN and CCN number concentrations peaked around 1200 LT (Fig. 4a). And the enhancement in bulk AR after 1200 LT was observed at higher \( SS \) because most of the particles were nucleation-mode particles, which were difficult to activate at lower \( SS \) (Fig. 4b). We examined jointly the changes in chemical composition (Fig. 4c) and size distribution (Fig. 4e), which showed that the increase in bulk AR until 1600 LT is largely attributed to the increase in sulfate and aged/oxidized organics. From 1200 to 1600 LT, the \( \kappa_{\text{chem}} \) and \( f_{44} \) increased simultaneously in Fig. 4d, while number concentration of accumulation-mode particles decreased.

Another peak in CCN number concentration and the bulk AR is seen from 0700 to 0800 LT. During this time, the mass fractions of nitrate, sulfate, and organics were all at a relatively stable level, but the number concentrations of aerosol particles with \( D > 100 \) nm increased significantly and the particle size showed a slight increasing trend (Fig. 4e). This means that the particle size took on a dominant role in controlling aerosol activity from 0700 to 0800 LT. Our analysis shows the integral effect of aerosol size and chemical composition on CCN activity over the entire period.

4.3. CCN closure study

Assuming internal mixing, a homogenous composition of the particles as provided by the ACSM, and simplified Köhler theory, we calculated the critical dry diameter \( (D_{\text{crit}}) \). The \( D_{\text{crit}} \) of a particle with properties \( (SS, \kappa_{\text{chem}}) \) corresponds to the minimum value of \( D \) needed for the particle to be activated. Particles with sizes larger than \( D_{\text{crit}} \) will be activated. The predicted total CCN number concentration \( (CN_{\text{Predicted}}) \) is obtained from the step-wise integration of the particle number size distribution \( (PSD) \) from \( D_{\text{crit}} \) to 600 nm, as 600 nm is the largest diameter in the processed data observed at the Xinzhou site. The time resolution of \( \kappa_{\text{chem}} \) derived from the ACSM is about 7.5 min and the time resolution of size-resolved CCN measurements is \( \sim 10 \) min. We assume that these two sets of measurements are simultaneous measurements if their drift time is within 5 min. Comparisons of predicted and measured \( N_{\text{CCN}} \) estimated using the derived \( D_{\text{crit}} \) for different \( SS \) levels are shown in Fig. 5.
Measured and predicted N_{CCN} at the different SS levels are moderately correlated (R^2 = 0.8–0.9). N_{CCN} was largely underestimated at SS equal to 0.075% and 0.17% (by 32% and 22%, respectively). The underestimation decreased as SS increased. The best agreement between predicted and observed N_{CCN} is seen when SS = 0.39% (the slope and R^2 are 0.99 and 0.86, respectively). At SS = 0.76%, CCN is overestimated by ~5%. Different performance of the CCN closure test under different supersaturations may be a result of differing chemical properties of different aerosol sizes. However, the large uncertainties in the prediction of CCN at lower SS are mainly caused by the less accuracy in setting lower supersaturation in CCN counter, as well as more noise in the concentrations at relatively larger sizes. Since we combined the SMPS with CCN counter, the measured CCN number concentrations were accordingly from a particle dry size range of 10–600 nm, the missing particles for which D_p > 600 nm should not be one of the reason to cause the uncertainty in CCN prediction at lower supersaturations in this study. These results show that the PSD measured by the SMPS, combined with the bulk k_{chem} calculated from the chemical composition information from the ACSM, can provide good estimates of N_{CCN} at higher supersaturations. Information about size-resolved chemical composition is expected a better agreement and more accurate estimates of CCN number concentrations. Improving the measurement accuracy at lower supersaturations may be important to the prediction of CCN.

4.4. Influence of NPF on CCN activity

To study the impact of NPF on CCN activity, two cases, one with an NPF event (24 August 2014) and another with a non-NPF event (4 August 2014), were selected for study. Simultaneously measured PSD, mass concentrations of organics, sulfate, nitrate, activation ratios, calculated k_{chem} and f_{44} were examined during the two events (Fig. 6).

During the NPF event, the PSD showed a “banana” shape representative of a typical NPF event. The NPF event spanned the period from 1000 LT to 1600 LT. The mass concentrations of sulfate and organics increased significantly (by a factor of 4.17 and 1.35, respectively) during the NPF event, indicating that particle growth may have been driven by the condensation of atmospheric sulfuric acid and volatile organics. Other field measurements have also demonstrated that sulfate and organics contribute to particle growth during an NPF event (Zhang et al., 2004b; Yue et al., 2011; Leng et al., 2014a; Minguillón et al., 2015). The evident spike in organic aerosol mass concentration prior to

Fig. 6. Particle size distribution during the NPF event (top three panels, 24 August) and the non-NPF event (bottom three panels, 4 August). The color contour in the top panel of each figure denotes dN/dlogD (cm^{-3}) on the left vertical axis (where N and D represent the particle number and size, respectively).
10:00 AM may be due to a local impact from traffic sources (Wang et al., 2016). The mass concentration of nitrate observed at Xinzhou increased slightly during the period of 1000 LT to 1600 LT. Simultaneously, $\kappa_{\text{chem}}$ and $f_{\text{det}}$ showed a slight increase during the period of 1000 LT to 1600 LT, indicating that particles were becoming more hygroscopic and oxidative. More hygroscopic and oxidative particles are expected to activate more easily.

At the beginning of the NPF, a burst of nucleation mode occurs, resulting in a very large number of particles; however most of the particles sizes were smaller than the critical diameter, for example, critical diameter for pure (NH$_4$)$_2$SO$_4$ particles is 83 nm at SS of 0.17%, thus activation fraction showed a significant drop. AR started to increase as the nanoparticles grew larger between 1100 LT and 1600 LT. The increase in AR started 1 h after the burst of nucleation mode particles. Overall, the NPF event enhanced CCN activity, especially at higher SS (e.g., ≥ 0.17%). The impact was insignificant when the SS was low (e.g., ≤ 0.13%) due to the limitation of particle critical diameters. In this case, the increase in $\kappa_{\text{chem}}$ and $f_{\text{det}}$ suggests that the particles were more hygroscopic, which would favor particle growth during the NPF event and enhancement of the AR.

During the non-NPF event (4 August 2014), average mass concentrations of sulfate, nitrate, and organics were 5.8 μg m$^{-3}$, 6.7 μg m$^{-3}$, and 2.4 μg m$^{-3}$, respectively. The calculated $\kappa_{\text{chem}}$ and $f_{\text{det}}$ changed slightly, suggesting an insignificant variation in particle hygroscopicity and oxidability. No “banana” shape was observed and accumulation-mode particles dominated all day long. The AR at all SS levels showed flat trends during the daytime. The AR decreased to <0.3 around 0600 LT and 1900 LT nucleation and accumulation mode peaks were seen in the PSD. The increase in small particles thus led to a reduction in AR.

To study the influence of the NPF event on CCN activation, simultaneous size-resolved CCN efficiency data measured during the periods of 1000–1200 LT and 1300–1500 LT, which were the key periods of NPF, were selected for further investigation (Fig. 7). From 1000 to 1200 LT, the aerosol number concentrations in the NPF case were enhanced dramatically in the nucleation and Aitken modes. We call this period the “growth” stage. From 1300 to 1500 LT, the particle size increased to ~100 nm and the particle number concentrations stopped increasing. We call this period the “leveling-off” stage. During the “growth” stage, the activation diameter ($D_a$) in the NPF case was on average larger than that observed during the non-NPF case. The increase in AR as $D_a$ increased observed during the NPF event was more gradual. This suggests that during the “growth” stage, particles in the NPF case were more heterogeneous with less hygroscopicity, so were more difficult to activate. In other words, the NPF during the “growth” stage restrained CCN activation. During the “leveling-off” stage, however, the situation changed. The CCN efficiency spectra during the NPF event showed smaller $D_a$, steeper increases in AR as $D_a$ increased. Therefore, the NPF during the “leveling-off” stage enhanced CCN activation. Based on our observations, NPF had different impacts on CCN activity at different times of the day during the NPF event. As shown in Fig. 7, the bulk AR from 1000 to 1200 LT was reduced during the NPF event, but began to increase after 1200 LT until reaching the accumulation mode size range.

5. Conclusions

In this study, we investigated the CCN activity and new particle formation impact on CCN activation based on the field campaign at the Xinzhou site in 2014. Both aerosol and CCN number concentrations were extremely high during the field campaign. The average bulk AR ranged from 12.6 ± 8% to 58.1 ± 15% for SS ranging from 0.075%–0.76%. We found that particle size was more related to the CCN activation ratios in the morning, whereas in the afternoon (~1400 LST), $\kappa_{\text{chem}}$ and $f_{\text{det}}$ were more closely associated with the bulk AR. Under the assumption of internal mixing, measured and predicted CCN

**Fig. 7.** Observed size-resolved CCN efficiency spectra for the NPF (red lines) and non-NPF cases (black lines) during the periods of 1000–1200 LT (top panel) and 1300–1500 LT (bottom panel). The different line thicknesses represent different SS levels.
number concentrations were compared using bulk chemical composition information based on κ-Köhler theory. We found that the predicted CCN number concentrations were underestimated by 20–30% at SS < 0.2% probably due to the measurement uncertainties. Estimates were more accurate at higher SS levels, suggesting that the hygroscopicity parameter based on bulk chemical composition information can provide a good estimate of CCN number concentrations. However, the size-resolved chemical composition is expected to improve the CCN number concentration prediction. The NPF showed a distinct effect on CCN activity at different periods. CCN activation was restrained at the growth stage, during which a larger size was required to reach the activation diameter, and accordingly bulk AR decreased by 500% at SS = 0.76%. CCN activation appeared to be enhanced during the “levelling-off” stage when the activation diameter was reduced and bulk AR began to increase quickly and was enhanced about 20% eventually. Our study showed that NPF events could either decrease or increase CCN number concentrations, as depends on their stage. According to our results, the effect or role of particles physiochemical properties on CCN activity largely depends on time of a day. In addition, it is important to classify the NPF events to at least two stages when evaluating the contribution of NPF to CCN number concentrations. Nevertheless, only one month data from one site is very limited to address the key issues in the research area and to parameterize CCN in models. Depending on the location, more field measurements and data are therefore necessary to validate the complex effect of particles properties on CCN activity, in particular in those heavily polluted regions.

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